# Determination the concentration of radon in human urine using LR-115 detector

# Essam M. Rasheed

Department of Physics, College of Science, Al-Nahrain University, Baghdad, Iraq

E-mail: esam esam777@yahoo.com

### Abstract

In the present study twenty samples of human urine were taken from healthy male and female with different of: ages, occupation and place of residence. These samples were collected from the hospital to measure the concentration of radon gas in human urine by using one of solid state nuclear track detectors LR-115.

The results obtained of the concentrations of radon in healthy human urine are varying from  $2.12\times10^{-3}$  Bq.l<sup>-1</sup> to  $4.42\times10^{-3}$  Bq.l<sup>-1</sup> and these values are less than the allowed limits  $12.3\times10^{-3}$  Bq.l<sup>-1</sup>.

# **Key words**

Urine, radon, LR-115, track detector, alpha particles.

#### Article info.

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# قياس تركيز الرادون في الإدرار البشري بأستخدام كاشف الأثر النووي 115-LR

قسم الفيزياء، كلية العلوم، جامعة النهرين، بغداد، العراق

### الخلاصة

استخدمت في هذه الدراسة عشرين نموذجا من الإدرار البشري والمأخوذة من أشخاص أصحاء لمختلف الاعمار والمهن ومكان العمل وجمعت هذه النماذج من المستشفى وقد قيس تركيز غاز الرادون لهذه النماذج باستخدام كاشف الأثر النووي LR-115.

وقد أظهرت النتائج التي تم الحصول عليها من خلال هذه الدراسة أن تراكيز الرادون المنبعثة من الإدرار البشري قد تراوحت من  $Bq.I^{-1}$   $Bq.I^{-1}$  إلى  $Bq.I^{-1}$   $Bq.I^{-1}$  وهذه القيم اقل من النسبة المسموح بها للأشخاص الأصحاء والبالغة  $Bq.I^{-1}$   $Bq.I^{-3}$ .

# Introduction

Radiation and radiation emitters (radionuclides) can expose the whole body (direct exposure) or expose tissue inside the body when inhaled or ingested. Different types of radiation vary in their ability to damage different kinds of tissue. All kinds of ionizing radiation can cause cancer and other effects. The main difference in the ability of alpha particles, beta particles, gamma-rays and x-rays to cause health effects is the amount of energy they have, their energy determines how far they can penetrate into tissue, it also

determines how much energy they are able to transmit directly or indirectly to tissues and the resulting damage [1].

The biological damage resulting from a given absorbed energy may be quite different for different tissues for these reasons, it is necessary to measure the radiation with an instrument and then translate it into response of tissue. Since different detectors do not have the same efficiency or sensitivity for all types of radiation and at all energies, there is no single instrument that can be used for alpha, beta and gamma [2].

The monitoring of trace metals in human urine plays an important role to gain a better understanding of sources of environmental contamination. However, human urine is a complex matrix containing high levels of urea, uric acid, proteins, fats, sodium,

potassium, bicarbonate and chloride[3].

# **Collection of the samples**

Twenty samples of human urine were taken from the hospital to measure the concentration of radon gas that emission from urine human, these samples are shown in Table 1.

Table 1: Urine human samples code.

Code of sample	Gender	Age (years)	Occupation	Place of residence
U1	male	32	worker	Baghdad
U2	male	44	worker	Baghdad
U3	male	45	worker	Basrah
U4	male	65	pensioned military man	Baghdad
U5	male	20	worker	Basrah
U6	male	22	farmer	Baghdad
U7	male	27	driver	Basrah
U8	male	17	student	Basrah
U9	male	40	driver	Baghdad
U10	male	66	pensioned officer	Baghdad
U11	female	23	student	Baghdad
U12	female	33	worker	Ramadi
U13	female	45	worker	Basrah
U14	female	03	housewife	Baghdad
U15	female	20	student	Baghdad
U16	female	35	worker	Basrah
U17	female	25	housewife	Baghdad
U18	female	29	housewife	Ramadi
U19	female	50	housewife	Ramadi
U20	female	55	housewife	Ramadi

# **Experimental details**

The human urine samples were collected about 0.1 litter and stored for one month at normal laboratory conditions which were kept in plastic containers. This time is necessary to get a radiological equilibrating to the samples, before determining the concentration of natural radioactive

material for the sample. Sheets of LR-115 were cut into small pieces each one of  $(1\times1\text{cm}^2)$  area, then we put the track detector LR-115 into inner cover side of cylindrical container for two months to registries  $\alpha$ -particles tracks that emitted from radon gas as shown in Fig1.

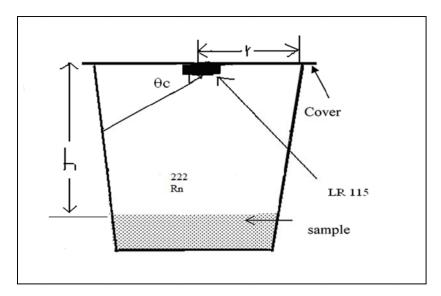


Fig.1: LR-115 detector with sample in the container.

After two months of the exposure, LR-115detectors were etched for (2.30) hr using chemical etching (NaOH) solution with (6.25N) at 55°C, the etched detectors were washed by

distilling water then dried and count the number of tracks under an optical microscope. The tracks in these samples that observed using LR-115 track detector as shown in Fig. 2.

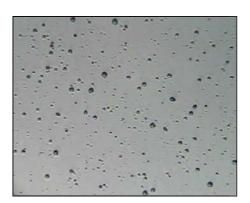


Fig.2: Image of LR-115 after etching.

## Results

The concentration of radium begins to increase with time (t) according to the Eq. (1) [4]:

$$C_{Rn} = C_{Ra} (1 - e^{-\lambda t})$$
 (1)

where  $\lambda$  is the decay constant of radon and  $C_{Ra}$  is the effective radium content of the sample (Bq.l<sup>-1</sup>), since the plastic track detector LR-115 measures the time integrated value of the above expression, the total number of alpha disintegrations in unit volume of tube with calibration factor K during the

exposure time (t), hence the track density is given by [5]:

$$\rho = (K * C_{Rn} * T_{eff})$$
 (2)

The radon concentration  $C_{Rn}$  of integrated radon exposure inside the tube was obtained from the track density of the detector, by using the calibration factor

 $K=(0.034\pm0.002)$ tracks.cm<sup>2</sup>/d.(Bq.m<sup>3</sup>) [6].

T<sub>eff</sub> which is the effective exposure time and equal to the flowing equation:

$$T_{\text{eff}} = t - \tau (1 - e^{-\lambda t}) \tag{3}$$

where  $\tau$  is the mean life of radon and equal to (5.5 days).

In this study the actual exposure time (t) is 60 days and the effective exposure time is:  $T_{eff} = 56.5$  days.

It is clear that the effective radium content of the sample can be calculated from the flowing formula: [7,8]

$$C_{Ra} = (\rho / K*T \text{ eff}) *(h*A/V)$$
 (4)

where A: the area of the cross – section of the tube in  $m^2$  and equal to  $(0.00785 \text{ m}^2)$ .

h: the distance between the detector and the top of the sample and equal to (0.17 m). V: the volume of the sample in liter and in this study equal to (0.1 liter).

All of these procedures were completed at normal condition without using heating or cooling. To determine the concentration of radon we used the Eq.(1).

The radon concentration in all samples (Bq.l<sup>-1</sup>) are shown in Table 2 and Fig.3.

Table 2: Radon concentration of urine human samples.

code of sample	$\rho$ tracks/cm <sup>2</sup>	C <sub>Rn</sub> ×10 -3 Bq.1-1
U1	722.85	2.58
U2	684.56	2.45
U3	920.63	3.30
U4	1020.74	3.66
U5	1232.75	4.42
U6	680.63	2.43
U7	950.34	3.40
U8	1100.22	3.95
U9	740.28	2.65
U10	890.66	3.20
U11	650.23	2.32
U12	876.24	3.15
U13	955.34	3.43
U14	590.66	2.12
U15	667.95	2.40
U16	1050.27	3.78
U17	780.44	2.80
U18	798.55	2.87
U19	840.20	3.02
U20	860.49	3.09

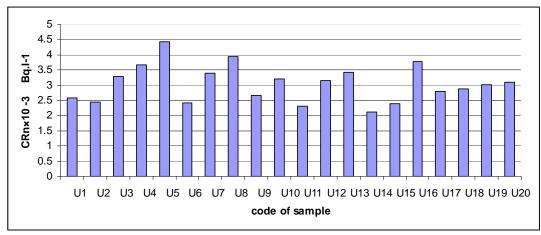


Fig.3: Radon concentration in all urine human samples.

The radon concentration in male urine samples are shown in Table 3 and Fig.4.

Table 3: Radon concentration in male urine human samples.

code of sample	$C_{Rn}^{\times}10^{-3}$ Bq.l <sup>-1</sup>
U1	2.58
U2	2.45
U3	3.30
U4	3.66
U5	4.42
U6	2.43
U7	3.40
U8	3.95
U9	2.65
U10	3.20
Average	3.204

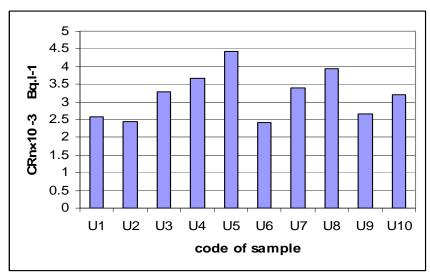


Fig. 4: Radon concentration of male urine human samples.

The radon concentration in female urine samples are shown in Table 4 and Fig.5.

code of sample	C <sub>Rn</sub> ×10 -3 Bq.l <sup>-1</sup>
U11	2.32
U12	3.15
U13	3.43
U14	2.12
U15	2.40
U16	3.78
U17	2.80
U18	2.87
U19	3.02
U20	3.09
Average	2.89

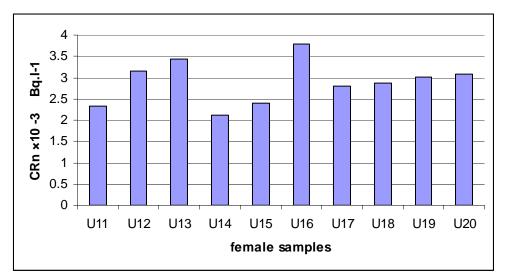


Fig. 5: Radon concentration of female urine human samples.

The difference between the average of radon concentration in male and female

urine samples are shown in Fig.6.

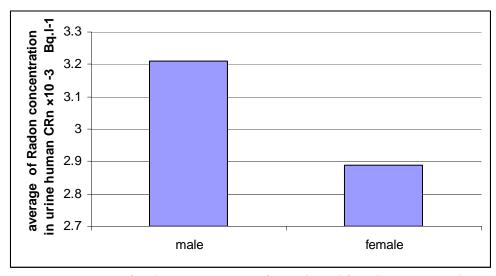


Fig.6: Average of radon concentration for male and female urine samples.

The radon concentrations in urine human of Baghdad samples are shown in Table 5.

The radon concentrations in urine human for Basrah samples are shown in Table 6.

Table 5: Radon concentration in urine human of Baghdad samples.

Code of sample	$C_{Ra} \times 10^{-3}$ Bq.l <sup>-1</sup>
U1	2.58
U2	2.45
U4	3.66
U6	2.43
U9	2.65
U10	3.20
U11	2.32
U14	2.12
U15	2.40
U17	2.80
Average	2.655

Table 6: Radon concentration in urine human of Basrah samples.

Code of sample	$C_{\rm Rn} \times 10^{-3}$ Bq.1 <sup>-1</sup>
U3	3.3
U5	4.42
U7	3.40
U8	3.95
U13	3.43
U16	3.78
Average	3.71

The radon concentrations in urine human for Ramadi samples are shown in Table 7. The radon concentrations

in urine human for Baghdad, Basrah and Ramadi samples are shown in Fig.7.

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Code of sample	$C_{Rn} \times 10^{-3}$ Bq.l <sup>-1</sup>	
U12	3.15	
U18	2.87	
U19	3.02	
U20	3.09	
Average	3.032	

Table 7: Radon concentration in urine human of Ramadi samples

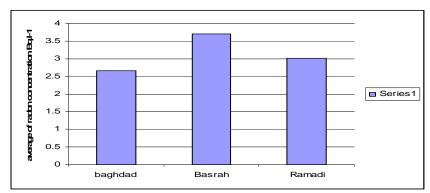


Fig.7: The average of radon concentrations in urine human for Baghdad, Basrah and Ramadi samples.

# **Discussion**

From Table 2, the radon concentration in all urine human samples vary from  $2.12 \times 10^{-3}$  Bq.I<sup>-1</sup> to  $4.42 \times 10^{-3}$  Bq.I<sup>-1</sup>, the minimum of radon concentration was in the sample U14- (female, 30year, housewife, Baghdad) and the maximum of radon concentration was in the sample U3- (male, 45year, worker, Basrah).

From Table 3, the radon concentration of urine human in male samples that vary from 2.43×10<sup>-3</sup> Bq.I<sup>-1</sup> to 4.42×10<sup>-3</sup> Bq.I<sup>-1</sup>, with the average of 3.21×10<sup>-3</sup> Bq.I<sup>-1</sup>. The minimum of radon concentration of urine human in sample U6- (meal, 22 year, farmer, Baghdad) and the maximum of radon concentration of urine human was in

sample U5- (male, 20year, worker, Basrah).

From Table 4, the radon concentration of urine human in female samples that vary from 2.12×10<sup>-3</sup> Bq.1<sup>-1</sup> to 3.78×10<sup>-3</sup> Bq.1<sup>-1</sup>, with the average of 2.89×10<sup>-3</sup> Bq.1<sup>-1</sup> The minimum of radon concentration of urine human in female was in sample U14-(female, 30years, housewife, Baghdad), the maximum of radon concentration of urine human in female was in sample U16-(female, 35 years, worker, Basrah).

From Table 3 and Table 4, we note that there is a difference of radon concentration in urine human between male and female samples because the radon concentration in urine human increased when the human is exposed for a long time to the radiation.

From Table 5 the radon concentrations of urine human in human samples where lived in Baghdad were  $2.12\times10^{-3}$ Bq.l<sup>-1</sup> to varied from  $3.66\times10^{-3}$ Bq.l<sup>-1</sup>, with the average of  $2.66\times10^{-3}$  Bq.l<sup>-1</sup>. The minimum of radon concentration of urine human was in sample U14- (female, 30 years, Baghdad) housewife. and maximum of radon concentration of urine human was in sample U4- (male, 65 years, pensioned military man. Baghdad).

From Table 6, the radon concentrations of urine human in human samples lived in Basrah where from  $3.30 \times 10^{-3}$  Bq.1<sup>-1</sup> to varied 4.42×10<sup>-3</sup> Bq.1<sup>-1</sup>, with the average of  $3.71 \times 10^{-3} \text{Bg.l}^{-1}$ . The minimum of radon concentration of urine human in sample U5-(male, 20years, was worker, Basrah) and the maximum of radon concentration of urine human was  $(4.42 \times 10^{-3} \text{Bq.l}^{-1})$  in sample U3-(male, 45 years, worker, Basrah).

From Table 7, the radon concentrations of urine human in human samples lived Ramadi where in from  $2.87 \times 10^{-3}$  Bq.1<sup>-1</sup> varied  $3.15\times10^{-3}$  Bq.l<sup>-1</sup>, with the average of  $3.01\times10^{-3}$  Bq.l<sup>-1</sup>. The minimum of radon concentration in human's urine was in sample U18- (female, 29years, housewife) and the maximum of radon concentration of urine human was in sample U12-(worker, 29 vears, housewife).

From Tables 5, 6 and 7, the average of radon concentration in humans urine from Baghdad samples is less than the radon concentration in human urine from Basrah samples because the radioactivity in Baghdad less than Basrah and the average of radon concentration of urine human from

Ramadi samples were between them because most areas of Ramadi contents a phosphate in soils and rocks.

#### **Conclusions**

From all the previous tables and results, the values of radon concentrations of urine human was dependence of the person's occupation, years of service, sex, age and health condition have been dealt with in sufficient details as required.

All these results are low compared with the allowed limit  $12.3 \times 10^{-3} \text{Bg.I}^{-1}$  [9].

#### References

[1] S.F. Hassan., "Determination of Uranium Concentration in Human Blood in Some Governorates of Iraq", M.Sc. Al- Nahrain University, 2006.

[2] A. F. A. Youssef and R. A. Farghli, Canad. J. Anal. Sci. Spec., 51 (2006) 288.

[3] J.Zhang, M. Chen, Ju, W. Z., Liu, S. J., Xu, M. J., Chu, J. H., Wu, T., J. Pharm. Biomed. Anal. 51 (2010) 685.

[4] P. Qu, Yan, S. C., Lu, H., Lu, Z. H., Microchim. Acta, 163 (2008) 321.

[5] B. X.Li, Guo, L. L., Xu, C. L., Ma, L. M., Spectrochim. Part, A 71 (2008) 892.

[6] A.A Ensafi, T.Khayamian, F.Hasanpour, J. Pharm. Biomed. Anal. 48 (2008) 321

[7] E.Nalewajko, A.Wiszowata, Kojło, A. J. Pharm. Biomed. Anal., 43 (2007) 210.

[8] B. Kozlowska, A. Walencik, J. Dorda, T. A. Przylibski , Radiat. Meas., 42 (2007) 1380.

[9] UNSCEAR "Sources, effects and risks of ionization radiation", United Nations Scientific Committee on the Effects of Atomic Radiation, Report to the General assembly, with Annexes, New York, p 34-40, 1993.